

Purpose: Royal Society workshop in “Unifying scientific disciplines to understand and solve emerging membrane filtration challenges”

Location: Chicheley, Newport Pagnell, Buckinghamshire, MK16 9JJ, UK

Dates: January 9-11, 2017

Hosts: Ian Griffiths, Davide Mattia, Sourav Mondal and Darrell Patterson

“Can we emulate Nature’s exquisite selectivity for water, ions and proteins?”

Georges Belfort

¹Howard P. Isermann Department of Chemical and Biological Engineering, and
Center for Biotechnology and Interdisciplinary Studies
Rensselaer Polytechnic Institute, Troy NY 12180-3590

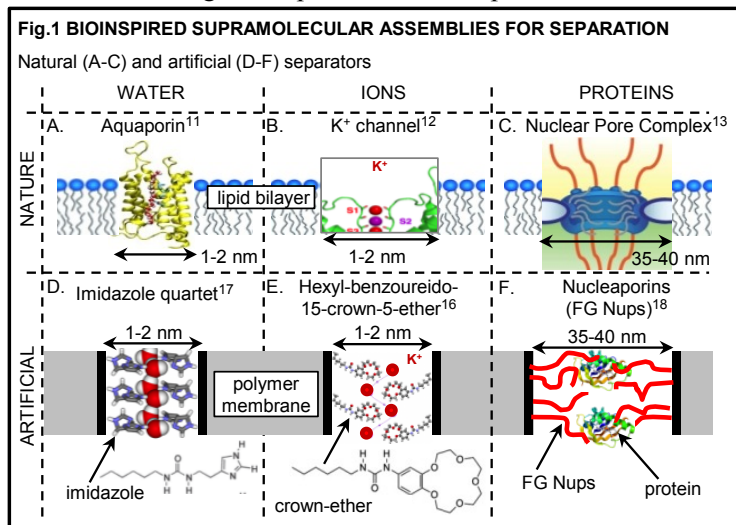
Creating a new class of membranes with the high selectivity of biological membranes and while maintaining large permeation fluxes is the holy grail of industrial membrane science and technology¹⁻⁵. Fundamental understanding of the mechanisms underlying biological membrane structure and function and incorporating that efficiently into synthetic materials would lead to a new class of membranes critical to the solution of several grand challenge problems including water purification and production of better and less expensive medicines⁶. For

example, biological membranes (containing phospholipid molecules and proteins) exhibit (a) exquisite selective transport of water, ions and proteins, among other molecules, at separation factors of $\alpha = 1$ as compared with synthetic membranes (i.e. $\sim 0.60 < \alpha < 0.985$ for bioprocessing and desalination of seawater)⁷, (b) but much lower permeation rates (i.e. $\sim 10^2$ - 10^3 lower for human Aquaporin (AQP) versus reverse osmosis) that is compensated for by very high surface areas in the body. They use transmembrane protein complexes (tmPCs) that comprise molecular-scale pores (Fig. 1A-C)⁸⁻¹³. The pores in these natural protein separators are exquisitely designed with chemical specificity, angstroms to nanometer length scales and with angstroms tolerances. Certain species like water or potassium ions permeate these structures while others, like sodium ions, are impermeable. Recent attempts to mimic some of the properties of biological membranes with artificial supramolecular assemblies (ASAs) highlight the magnitude of our knowledge gap (Fig. 1D-F)¹⁴⁻¹⁸. Their permeation rates are in some cases orders of magnitude worse than corresponding natural systems.

In this presentation, we will compare different approaches for synthesizing nature-inspired membrane separators with separation factors of $\alpha \sim 1$, describe progress to date, and present current challenges.

References

1. Geise GM, Lee HS, Miller DJ, Freeman BD, McGrath JE, Paul DR. Water Purification by Membranes: The Role of Polymer Science. *Journal of Polymer Science Part B-Polymer Physics*. 2010;48(15):1685-1718.
2. Greenlee LF, Lawler DF, Freeman BD, Marrot B, Moulin P. Reverse osmosis desalination: Water sources, technology, and today's challenges. *Water Research*. 2009;43(9):2317-2348.
3. La YH, Diep J, Al-Rasheed R, et al. Enhanced desalination performance of polyamide bi-layer membranes prepared by sequential interfacial polymerization. *Journal of Membrane Science*. 2013;437:33-39.
4. Merkel TC, Freeman BD, Spontak RJ, et al. Ultraporous, reverse-selective nanocomposite membranes. *Science*. 2002;296(5567):519-522.



5. Merkel TC, Freeman BD, Spontak RJ, et al. Sorption, transport, and structural evidence for enhanced free volume in poly(4-methyl-2-pentyne)/fumed silica nanocomposite membranes. *Chemistry of Materials*. 2003;15(1):109-123.
6. NAE Grand Challenges for Engineering. The National Academies of Science, Engineering and Medicine; 2016.
7. Baker RW. *Membrane Technology and Applications*. 3 ed: John Wiley & Sons, Ltd; 2012.
8. Agre P, Bonhivers M, Borgnia MJ. The aquaporins, blueprints for cellular plumbing systems. *J Biol Chem*. 1998;273(24):14659-14662.
9. Doyle DA, Morais Cabral J, Pfuetzner RA, et al. The structure of the potassium channel: molecular basis of K⁺ conduction and selectivity. *Science*. 1998;280(5360):69-77.
10. Lin DH, Stuwe T, Schilbach S, et al. Architecture of the symmetric core of the nuclear pore. *Science*. 2016;352(6283):aaf1015.
11. Agre P. Aquaporin water channels (Nobel lecture). *Angewandte Chemie-International Edition*. 2004;43(33):4278-4290.
12. MacKinnon R. Potassium channels and the atomic basis of selective ion conduction (Nobel lecture). *Angew Chem Int Edit*. 2004;43(33):4265-4277.
13. Grunwald D, Singer RH, Rout M. Nuclear export dynamics of RNA-protein complexes. *Nature*. 2011;475(7356):333-341.
14. Barboiu M. Artificial water channels - incipient innovative developments. *Chem Commun (Camb)*. 2016;52(33):5657-5665.
15. Si W, Chen L, Hu XB, et al. Selective artificial transmembrane channels for protons by formation of water wires. *Angew Chem Int Ed Engl*. 2011;50(52):12564-12568.
16. Gilles A, Barboiu M. Highly Selective Artificial K⁽⁺⁾ Channels: An Example of Selectivity-Induced Transmembrane Potential. *J Am Chem Soc*. 2016;138(1):426-432.
17. Licsandru E, Kocsis I, Shen YX, et al. Salt-Excluding Artificial Water Channels Exhibiting Enhanced Dipolar Water and Proton Translocation. *J Am Chem Soc*. 2016;138(16):5403-5409.
18. Jovanovic-Taliman T, Tetenbaum-Novatt J, McKenney AS, et al. Artificial nanopores that mimic the transport selectivity of the nuclear pore complex. *Nature*. 2009;457(7232):1023-1027.